



STUDY ON SYNTHESIZING BIODIESEL CATALYZED BY SOLID BASE CATALYST

S.F. Li, H.B. Ma, B.Y. Wang, R.H. Wang, S.J. Tian

Key Laboratory for Green Chemical Technology of State Education Ministry, School of Chemical Engineering and Technology, Tianjin University, Tianjin, China

shfli@tju.edu.cn

Biodiesel is being developed as one of the most suitable substitutions of fossil fuels in solving energy problems as well as the relative environmental problems. In this work two kinds of solid bases were studied for producing biodiesel with rapeseed oil and methanol which were considered to be environmental benign process.

The solid base catalyst $\text{KOH}/\gamma\text{-Al}_2\text{O}_3$ was prepared by impregnation method. The experimental results demonstrated that the catalyst $\text{KOH}/\gamma\text{-Al}_2\text{O}_3$ has high activity under the optimized preparation conditions. With the prepared $\text{KOH}/\gamma\text{-Al}_2\text{O}_3$, the biodiesel yield can reach as high as 84.20%. The prepared catalyst was characterized by Hammett indicator method, CO_2 -TPD, XRD, SEM, BET, IR and XPS. It can be found that the Al-O-K species is produced and both the basic strength and the basicity of the catalyst are increased after $\gamma\text{-Al}_2\text{O}_3$ is loaded with KOH upon calcinations. Moreover, the heterogeneous base $\text{KOH}/\gamma\text{-Al}_2\text{O}_3$ is more tolerant to free fatty acid and water contents compared to homogeneous KOH.

The solid base $\text{K}/\text{KOH}/\gamma\text{-Al}_2\text{O}_3$ was also prepared by impregnation followed with high temperature reaction. It was firstly used in catalyzing the transesterification of rapeseed oil with methanol to synthesize biodiesel. Both the catalyst preparation conditions and the reaction conditions of the transesterification were optimized. The results demonstrated that the biodiesel yield can reach as high as 84.52% with the prepared $\text{K}/\text{KOH}/\gamma\text{-Al}_2\text{O}_3$. The characterization results of catalyst show that after metal potassium is subsequently loaded onto $\text{KOH}/\gamma\text{-Al}_2\text{O}_3$ to form catalyst $\text{K}/\text{KOH}/\gamma\text{-Al}_2\text{O}_3$, not only the Al-O-K species increase, but also the surface color centers of $\text{F}_s^{(+)}$ are produced. And thus, the surface of catalyst $\text{K}/\text{KOH}/\gamma\text{-Al}_2\text{O}_3$ has stronger electron donor sites. It was also found that only 4.2% potassium was extracted from catalyst $\text{K}/\text{KOH}/\gamma\text{-Al}_2\text{O}_3$ after the first time use. With the recycled catalyst, the biodiesel yield reached 75.61%, suggesting that the catalyst has relatively high chemical stability.